

**The Influence of Purity Level on
the Mechanical Properties of Hot
Isostatically Pressed Beryllium**

B. C. Odgaard, Jr.



Sandia Laboratories
energy report



U.S. GOVERNMENT PRINTING OFFICE
WASHINGTON, D. C. 20540
1964 O - 348-100

MASTER

THE UNIVERSITY OF CALIFORNIA LIBRARY
400 CHASE DRIVE
BERKELEY, CALIFORNIA 94720

Dr. F. Oswald, Jr.
Materials Development Division II
Sandia Laboratories
Livermore, California 94550

ABSTRACT

The procurement of a quantity of ultra-pure beryllium powder combined with special handling from powder to billet form resulted in the fabrication of high purity beryllium. The mechanical properties of these billets were contrasted to those of commercial grade billets to determine the influence of impurities and powder processing.

The tensile test results show that the strength values are primarily dependent on the grain size in a behavior predictable by the Hall-Petch relationship. Only a fraction of the strength differential can be attributed to metallic impurities in solution. The grain size is controlled by the powder size distribution. The ductility is dominated by both grain size and oxide content. The fine grained, low oxide billets exhibited the highest ductilities. There is evidence to suggest that oxide distribution has a large influence on the ductility.

The fracture toughness was highest for the high purity beryllium billets.

REFERENCES

The author acknowledges the help and assistance of Dr. T. H. London, University of Maryland, and Dr. J. H. Van Dyke, University of Michigan, in the development of the work described in this paper. The author also acknowledges the help and cooperation of Paul J. Combs, Lowell Kays, and J. W. McCall in the simple expansion and heating.

*Dr. London is now with the Naval Air Development Center Warminster, Pa.

THE INFLUENCE OF PURITY LEVEL ON THE MECHANICAL
PROPERTIES OF HPT ISOSTATICALLY PRESSED BERYLLIUM

Introduction

Several studies have preceded this work relating the influence of grain size, oxide content and total chemistry to the room temperature mechanical properties of beryllium.[1-4] This study utilized high purity powders of varying compositions and inert gas handling at the Kaweco Berylco facility to fabricate beryllium billets. The result was a fine grain, low impurity material which is difficult to produce. Tests were conducted on this material and contrasted to tests on three commercially available grades of beryllium. This is reflected in the test matrix shown in Figure 1, which enable the study of powder grain size, chemistry and processing history on the microstructural and mechanical properties.

Grade	Grain Size		
	100	200	400
99.99%	100-1000	100-200	100-400
99.95%	100-1000	100-200	100-400
99.90%	100-1000	100-200	100-400
99.85%	100-1000	100-200	100-400
99.80%	100-1000	100-200	100-400
99.75%	100-1000	100-200	100-400
99.70%	100-1000	100-200	100-400
99.65%	100-1000	100-200	100-400
99.60%	100-1000	100-200	100-400
99.55%	100-1000	100-200	100-400
99.50%	100-1000	100-200	100-400
99.45%	100-1000	100-200	100-400
99.40%	100-1000	100-200	100-400
99.35%	100-1000	100-200	100-400
99.30%	100-1000	100-200	100-400
99.25%	100-1000	100-200	100-400
99.20%	100-1000	100-200	100-400
99.15%	100-1000	100-200	100-400
99.10%	100-1000	100-200	100-400
99.05%	100-1000	100-200	100-400
99.00%	100-1000	100-200	100-400

Figure 1. Program Matrix - showing the billet identification numbers for each particle size and grade of powder.

Material:

The materials used in this investigation were produced by Kawecki Berylco Industries using a hot isostatic pressing process. The pressing schedule offered the advantage of consolidation at much higher pressures (103 MPa) over conventional vacuum hot pressing and thus lower temperatures to accomplish full densification. The low temperatures minimize grain growth and the isostatic pressure produces a near isotropic material. The high pressures also allow the consolidation of high purity powders to full density. Billet compositions are shown in Table 1.

Table 1
BILLET COMPOSITION

Billet No.	Grade	Particle Size μm	BeO ¹ %	C ² ppm	Fe ³ ppm	Al ppm	Mg ⁴ ppm	Ni ⁵ ppm	Mn ⁶ ppm	Cr ⁷ ppm	Co ⁸ ppm	Cu ⁹ ppm	Si ppm	Ti ¹⁰ ppm
301	P-10	-44	1.02	680	1090	410 ⁸	260	160	85	95	<5	75	190 ⁹	190
303	*	-44+10	.64	560	1000	340 ⁸	270	135	90	115	<5	70	175 ⁹	125
305	*	-20	1.89	400	1350	560 ⁸	385	135	95	115	<5	70	180 ⁹	280
307	SP-10	-44	.53	270	950	100 ⁸	50	130	60	60	<5	65	130 ⁹	300
310	*	-44+10	.43	300	950	100 ⁸	50	110	55	50	<5	60	130 ⁹	.55
311	*	-20	1.00	300	1000	110 ⁸	45	110	50	90	<5	55	175 ⁹	135
314	P-1	-44	.95	180	170	<20 ⁸	5	50	3	10	50	15	31 ⁹	<10
315	*	-44+10	.52	180	140	<20 ⁸	4	110	<3	<5	9	10	46 ⁹	<10
317	*	-20	1.52	300	260	25 ⁸	8	105	8	8	25	21	45 ⁹	<10
318	P-0	-44	.43	M.D.	70	45 ⁸	5	3	7	.5	<5	15	40 ⁹	<10
321	*	-44+10	<.01	250	120	30 ⁸	12	20	12	<5	8	20	50 ⁹	<10
322	*	-20	1.16	250	810	35 ⁸	28	25	17	7	8	25	50 ⁹	<10

Notes: ¹ Neutron Activation
² Conductometric
³ Atomic Absorption
⁴ Spectrographic
⁵ Wet Chemistry

Powders:

The P-10 powder was derived from magnesium reduced bead by vacuum melting, casting into small ingot form, lathe turning to chips and impact attritioning.

The latter is a process by which chips are fed into a hopper and carried by high velocity, cold, inert gas to a beryllium target. The impact velocity is sufficient to fracture the chips into smaller particles. The result is a powder of varying sizes ranging from sub-micron to >44 μm . Subsequent separation into size ranges (powder classification) produced the fractions used in this study: -44+10 μm , -44 μm , and -20 μm . The XP-10 grade was produced by further processing the P-10 powder using an acid leaching step to reduce impurities. As noted in Table I, the BeO, Al, Mg and to some extent C, Mn and Cr are lowered by this process.

The P-1 powder was processed from electrolytic flake (grade EF-1) and the P-0 powder from twice electrolyzed flake (EF-0). This step utilizes an electrolytic process to reduce the impurity content of the magnesium reduced bead. The BeO content was minimized by processing the beryllium in an argon environment whenever possible. The final comminution was accomplished in a beryllium-lined ball mill (tungsten carbide balls) filled with argon. Powder classification into the ranges discussed earlier was done in argon.

Consolidation:

The flow diagrams for powder processing to billet form are shown in Figures 2 and 3. The P-10 and XP-10 powders were hot isostatically pressed at a pressure of 103 MPa (15 ksi) at 915°C; the P-1 and P-0 powders at 103 MPa and 1065°C. The higher temperature used to consolidate the latter powders was to ensure full densification. The XP-10 billets were given an additional heat treatment (750°C, 2 hours, air cool) to eliminate the low ductility exhibited in the as-pressed condition.

The schedule used in powder consolidation was to (1) load the powders into a stainless steel cylindrical can, (2) weld the end closed with an attached stem, (3) vacuum bake out the can, (4) pinch off the stem and (5)

process the assembly in a HIP chamber at a predetermined pressure, temperature and time schedule.

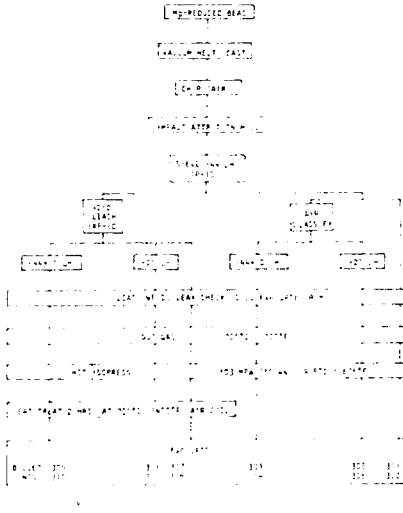


FIGURE 2 FLOW DIAGRAM FOR P-10 AND P-11 MATERIAL

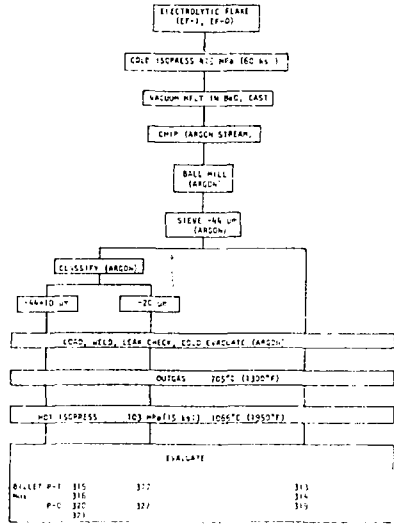


FIGURE 3 FLOW DIAGRAM FOR P-10 AND P-11 MATERIAL

Experimental Technique:

Tensile tests were conducted on an Instron machine on specimens having a 12.7 mm gage length and 3.56 mm diameter; at a cross head velocity of 8.5×10^{-4} cm/sec. Strain was measured using a 7.62 mm extensometer attached to the specimen. Surface damage from machining was removed by etching 0.08 mm from the surface of all test specimens.

Compressive test data was generated on an Instron machine with cylindrical samples 6.35 mm in length and a 5.00 mm diameter.

Toughness tests were accomplished using a WOL geometry with a thickness of 9.91 mm. The cross head velocity for all tests was 8.5×10^{-4} cm/sec.

(Ref. ASTM-E399).

The toughness data is shown in Table III. The K_{IC} value is derived by the equation

$$K_{IC} = PS/BW^{3/2} f \left(\frac{a}{W} \right) \quad (\text{Ref. ASTM-E399})$$

where P = load, B = specimen thickness,

S = span, W = specimen depth and

a = crack length.

Table III TOUGHNESS DATA		
Billet No.	Material/ Particle Size µm	K_{IC} MPa-m ^{3/2}
301	P-10/-44	11.65
303	P-10/-44+10	10.85
305	P-10/-20	9.93
307	XP-10/-44	8.98
309	XP-10/-44+10	14.29
311	XP-10/-20	11.24
314	P-1/-44	15.11
315	P-1/-44+10	14.46
317	P-1/-20	11.47
319	P-0/-44	14.56
321	P-0/-44+10	17.90
322	P-0/-20	16.28

Discussion:

Strength -- The yield point phenomenon, indicated by the value, σ_{y1} (Table II), for the P-10 and XP-10 grades, has been suggested as due to the metallic impurity, iron.[6] Heat treating at 1023K for 5 hours followed by slow cooling (25°C per hour) to 523K produces a precipitate, removes the iron from solution, and eliminates the yield point. No work was done in this study to dispute or

support this claim. However it is noteworthy that only the high metallic impurity grades exhibited this phenomenon.

The tensile yield strengths of the P-10 and XP-10 grades are contrasted to those of the P-1 and P-0 grades as influenced by grain size in Figure 4. The strength dependence on grain size is predicted by the Hall-Petch relationship. The higher yield strengths on the P-10 and XP-10 grades are thought to be due to the solid-solution strengthening effect of the metallic impurities. This theory is supported by Figure 5, where the grain size dependence has been normalized and the higher impurity billets (P-10, XP-10) are moderately higher in strength. The strength differential is about 100 MPa.

The compressive yield strengths follow the same behavior as in tension but at a consistently lower value indicating a degree of anisotropy.

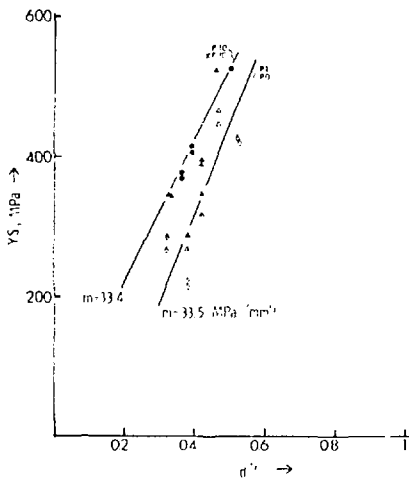


Figure 4 - The influence of grain size on the yield strength.

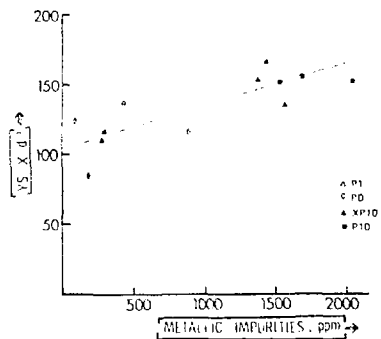


Figure 5 - The influence of metallic impurities on the yield strength (normalized for grain size).

Ductility:

The ductility within a given beryllium grade was influenced strongly by the powder input size; the finest size powder (smallest grain size) consistently yielded the lowest ductility. To infer that grain size is singly affecting the ductility value is an over simplification as the impurity level for a given grade of beryllium goes up as the powder size goes down. The increase in impurities may contribute to the reduction in ductility. Each grade was capable of yielding a high ductility (0.04 strain to failure). In recent studies by Aldinger et al. on similarly processed beryllium grades including F-1, P-10, RR 243* and cast material, it was shown that elongation was dependent on grain size and beryllium oxide content. Figure 6 shows the elongation (normalized for grain size) as influenced by beryllium oxide content. The

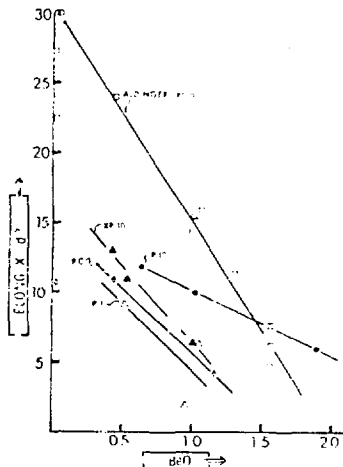


Figure 6 - The influence of beryllium oxide on elongation (normalized for grain size).

*Brush-Wellman Product

of oxide sizes with isolated regions where the oxides had agglomerated. This is illustrated in Figures 7 and 8.

The P-1, P-3 and X-10 grades are inherently lower in impurities and oxides because of the methods used in processing the powders. The test results indicate that the microstructural differences between these grades and the 10 grade are great enough to effect the ductility and toughness behavior.

CONCLUSIONS

1. The tensile strength and elongation of the test specimens were directly proportional to the ductility levels regardless of the impurities and oxides content.

2. The ductility of the test specimens was lower for the 10 grade than for the P-1 grade which consistently produced higher tensile strength and the lowest ductility.

3. The ductility and elongation of the test specimens was controlled by the grain size and not by the oxide content or impurities. The impurity content of the test specimens had no effect on ductility or elongation over the 10 to 100 ppm range of impurities examined.

4. The ductility was directly proportional to oxide content, with the lowest ductility for the 10 grade.

5. The ductility of the test specimens was highest for the near purity grade billets. The lowest ductility was exhibited by billet #301 which had the lowest ductility oxide content.



Figure 1. TEM images of CdTe nanowires prepared by the wet chemical method. The diameter of the left is about 0.5 μm and the right is about 1 μm .



Figure 2. TEM images of CdTe nanowires prepared by the wet chemical method. The diameter of the left is about 0.5 μm and the right is about 1 μm .

References

1. V. Ye Ivanov et al, "Influence of Grain Size on the Cold Shortness of Beryllium", *Physics of Metals and Metallography*, 31, 1971, 158.
2. V. Ye Ivanov et al, "Influence of Impurities on the Cold Shortness of Beryllium", *Physics of Metals and Metallography*, 31, 1971, 164.
3. H. Goual, I. Perlmutter, Beryllium as a Technological Material, presented in the proceedings of the Grenoble Conference on Beryllium, May 17-20, 1965.
4. W. W. Beaman, V. G. Gulyaev, Influence of Grain Size-Purity Relationships and the Factors that Affect Beryllium and the Resulting Mechanical Properties, 1971.
5. A. W. Thompson, "Effect of Grain Size on the Cold Shortness", *Metallography* 6, 1972, 304-311.
6. H. G. Goual, "Effect of Yield Point Phenomena in Commercial Beryllium on Fatigue", *Metallurgische Arbeiten* #33-230, 196, 1, 1974.