

ION-BEAM MIXING OF CERAMIC ALLOYS; PREPARATION AND MECHANICAL PROPERTIES*

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ABSTRACT

Techniques used to produce unique states of pure metals mixed into ceramic materials are presented. The samples were prepared by irradiating a 1-MeV Fe⁺ beam on Al₂O₃ crystal surfaces over which a thin chromium or zirconium film had been evaporated. The limitations of using noble gas ion beams are noted. Micro Knoop hardness tests performed near the surfaces of the samples indicated a significant increase in the hardness of most samples prepared by ion beam mixing.

INTRODUCTION

One of the results of ion implantation studies carried out over the past few years has been the recognition that ion beams induce interactions between atoms even at temperatures well below thermal activation temperatures. Such ion processing techniques as ion-beam mixing (IBM) initiate atomic interactions leading to equilibrium and metastable surface alloys as well as amorphous materials. As ions pass through interfaces of, for example, two distinct chemical regions in a sample, some energy is deposited by these primary ions in a region surrounding the interface. The combination of the ionization, lattice vibrations, and Frenkel pairs produced by the primary ion stimulate atomic migration across the interfacial boundary. Some atoms experience direct collision with the primary ions, and this process is referred to as "recoil implantation" [1], whereas the atomic motion associated with the radiation cascades is referred to as "cascade mixing" [2-4]. The atomic cascades can initiate large concentration gradients of both interstitials and vacancies, which enhance diffusion [5-6], lead to solute migration [7] and phase transformations [8], or dissolution of precipitates [9]. A review of some of these processes has recently been given by Myers [10]. Final states produced in this way may be unique low energy states or new metastable phases. Reference [11] contains a review covering ion implantation studies outside of semiconductor technology with special emphasis on material properties.

Ceramics have a special importance in modern technology. In addition to refractory fissionable fuels, the electrical properties of ceramics will see broad application in fusion reactors such as first wall coatings, blanket gaps, injectors, divertors, limiters, magnetic insulation, and windows. These applications require a high degree of mechanical stability.

In this work we are continuing an investigation of the changes in mechanical properties of ceramics due to ion beam irradiation [12]. In particular, we have examined the hardness changes near the surface of Al₂O₃ when chromium or zirconium are introduced by the mechanism of ion beam mixing.

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Ion bombardment of Al_2O_3 has been carried out by Drigo et al. and Carnera et al. [13-14] who determined lattice disorder and atom locations created by implanting 20-100 keV Pb^+ ions. Other investigators have studied defect structure produced by the implantation of H, He, Ar, Kr, and Xe as well as self ion implantation of aluminum and oxygen [15-20]. Although no data on mechanical properties have been reported, a group at Sandia National Laboratory has calculated surface stress produced in Al_2O_3 from a measurement of the deflection of a cantilever beam specimen during implantation [21].

MATERIALS AND METHODS

Single crystals of Al_2O_3 were obtained from two sources, Union Carbide Corporation (Crystal Products Division, 888 Balboa Avenue, San Diego, CA 92123) and Crystal Systems, Inc. (P.O. Box 1957, Salem, MA 01970). Both materials had high-purity, low-dislocation density and were oriented within ± 2 of (0001). Before implantation, each specimen was annealed for five days at 1200°C in air to remove any mechanical damage introduced during sample preparation.

Chromium or zirconium films of $\approx 160 \text{ \AA}$ were evaporated on the (0001) surface of most of the crystals. The remaining crystals were irradiated without the evaporated layer. Irradiations were carried out with 1 MeV Fe^+ ions in a UHV target chamber at a pressure of 10^{-8} torr. The ion beam was normal to the crystal surface and limited to maintain a specimen temperature $\lesssim 300^\circ\text{C}$. Inert gas ions were not used because surface blisters formed at high fluence. Following the irradiation, the excess metal film was etched away with hot potassium permanganate.

Rutherford backscattering and channeling (0001 axial) techniques were applied to the irradiated targets in order to characterize the atomic migration and radiation damage. The energy of the backscattered ^4He particles was measured by a silicon surface barrier detector with energy resolution 15 keV (FWHM) positioned 140° with respect to the incident beam. The beam energy was 1.6 or 0.5 MeV in the case of the channeling experiments. The beam current was limited to a few nanoamps to minimize pulse pileup distortion of the spectra. An electronic pileup rejector was also used. Details of simple film targets such as gold evaporated on Si were examined in order to characterize the remaining pile-up effects and the ion beam degradation by finite apertures. This procedure increased our certainty of the magnitude of the metal ion concentrations in regions where the magnitude is only of the order of 100 appm. The actual depth profiles were calculated from the backscattering spectra by direct deconvolution [22]. The data were normalized by the Al component in Al_2O_3 substrate backscattering.

Mechanical properties were evaluated by microhardness measurements. In all cases, only half of a specimen surface was implanted so that property measurements of unimplanted and implanted regions could be made under identical conditions. The hardness properties will be presented as relative values, i.e., ratio of implanted value to unimplanted value. The Knoop hardness technique was used with a force of 0.147 N in order to confine the impression to the near surface region. Typical impression depths were about 3000 \AA .

RESULTS

The depth profile measurement for the Cr film on Al_2O_3 is given in Fig. 1. The range of the Fe^+ ions is about 4000 \AA and is mostly off scale. Also shown in the figure is a profile following a thermal anneal for 3 h at 600°C but without any ion irradiation (except for the ^4He analysis which results in orders of magnitude less dpa than the Fe irradiation). The background profile is one taken on a virgin Al_2O_3 sample. The ion mixed profile is quite

different from the annealed-only in two ways: (a) the distinct broadening of the near surface $\approx 500 \text{ \AA}$ profile and (b) a much larger and nearly featureless concentration in the depth region $>500 \text{ \AA}$. A similar measurement on Fe irradiated Zr is shown in Fig. 2. These results differ from the Cr case only in the deeper minimum around 1200 \AA . Finally, it was noted in all irradiations, crystals darkened. By etching the surface it was found that the discoloration occurred throughout the depth of the implant to approximately 1 \mu m

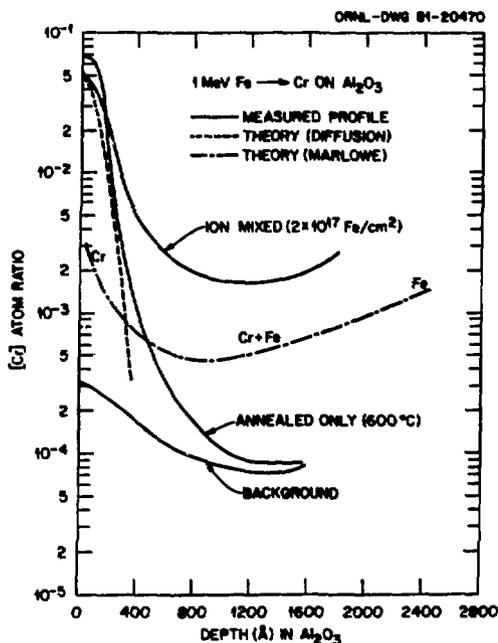


Fig. 1. Depth-concentration profiles of Cr coated Al_2O_3 samples following Fe^+ ion irradiation at $T < 300^\circ\text{C}$.

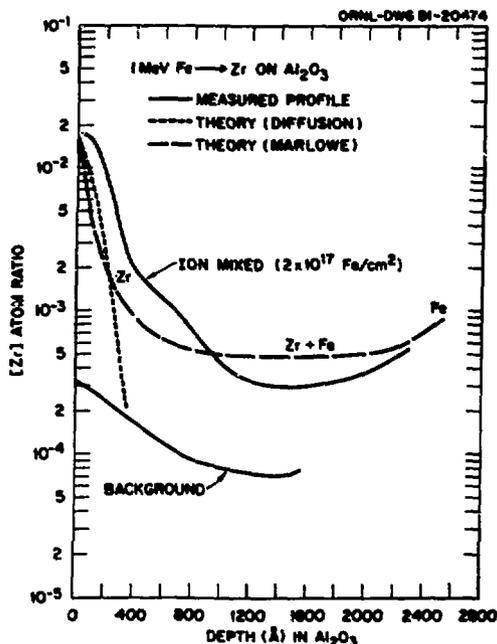


Fig. 2. Depth concentration profiles of a Zr coated Al_2O_3 sample following Fe^+ ion irradiation at $T < 300^\circ\text{C}$.

Channeling spectra corresponding to the profiles in Figs. 1 and 2 are given in Fig. 3 for Cr and Zr. The near surface region in Fig. 1 corresponds to channels near 250 for Cr and 125 for Al backscattering in Fig. 3. The yield (counts/channel) in both surface regions for the aligned spectrum are much lower than in the random spectrum. In addition, the Al backscattering from the deeper regions of the crystal corresponding to channels below 100 shows little difference between the aligned (irradiated) curve and the virgin (unirradiated) case. A similar relationship can be seen in Fig. 3 for the analysis of the irradiation of the Zr film. Quantitative values are given in Table I.

The χ_{Al} (channeling minima) values shown in Table I were used along with χ_{Cr} and χ_{Zr} minima in order to estimate the fraction of substitutionality of the ion mixed metal atoms. These results are given in the right-hand column of Table I. The result of the hardness tests indicated that the hardness increased 21% for the Cr mixed surface and 28% for the Zr mixed case. An irradiated surface without evaporated film showed a hardness increase of 15%.

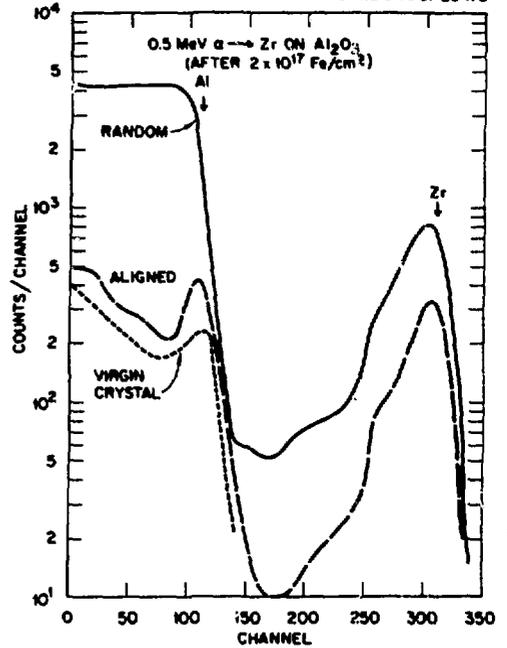
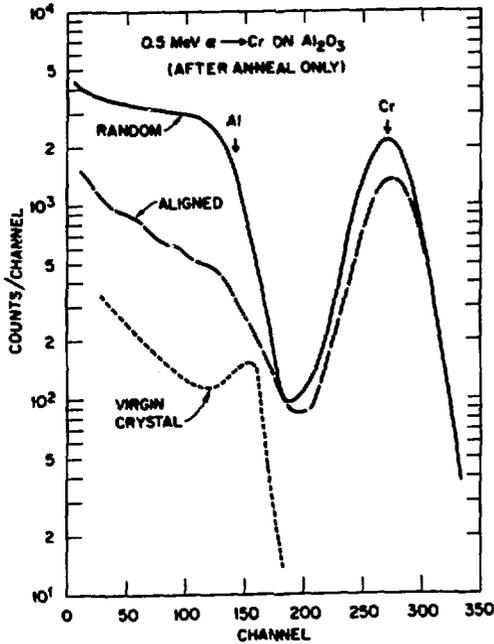


Fig. 3. Rutherford backscattering channeling (0001) spectra corresponding to the Cr (left) sample in Fig. 1 and the Zr (right) sample in Fig. 2.

TABLE I

A summary of the results of the (0001) channeling measurements. In column 3, χ refers to the minimum yield for Al in the substrate; in column 4, f_s is the fraction of metal film atoms found in substitutional type sites which would be consistent with the results of the Cr, Zr, and Al channeling minima.

Film	Fluence (Fe/cm ²)	χ_{Al}	f_s (= $\frac{1 - \chi_M}{1 - \chi_{Al}}$ %)
Cr	2×10^{17}	0.04	85
Zr	2×10^{17}	0.04	66
Zr	4×10^{17}	0.14	58

DISCUSSION

Currently, the degree of lattice disorder in Al₂O₃ per unit fluence appears much less than in previous studies [13-21]. For example, the increase in the aligned spectra minima for the 40 keV Kr bombardment [20] occurs at about two orders of magnitude lower dose (10^{15} /cm²) than our 1 MeV Fe irradiations (10^{17} /cm²). In these previous studies, the ranges of the implanted ions were much shorter than the present experiments. Thus the surface disorder caused by the presence of ions themselves (as opposed to the atomic displacements

created by the ions) may be largely responsible for the dechanneling or back-scattering in the aligned spectra. Using the energy deposition code E-DEP [23], we investigated this quantitatively by calculating the displacements (dpa) in specific cases. The results indicated that the near surface region displacement damage was about 2 dpa for 10^{15} Kr/cm², but was about 70 dpa for 2×10^{17} Fe/cm². An additional reason for the low lattice disorder in the present case is that the ratio of ionization to atomic displacement events is relatively large for the 1 MeV beam in the near surface regions. There is evidence to suggest that such ionization processes anneal displacement damage in Al₂O₃ near room temperature [15,16].

The fraction of substitutional Cr or Zr atoms resulting from ion beam mixing is more than half, even for a beam fluence of 4×10^{17} Fe/cm². This is consistent with the results of Drigo [13] who found a high degree of Al substitutional site occupancy for implanted Pb ions. The degree of substitutionality is somewhat greater for Cr than Zr and is fluence dependent for the Zr in the 10^{17} /cm² range.

Another aspect of using higher energy beams in ion beam mixing experiments is the larger contribution of recoiling ions. This is shown in Figs. 1 and 2 where calculations of migration profiles are compared. The diffusion curve is based upon the simple assumption [10] that the temperature independent diffusion constant is

$$D = \frac{1}{6} \lambda^2 P, \quad (1)$$

where the jump distance λ is taken as 20 Å, and P is given by the Kinchin-Pease relation in which the energy deposition is calculated by E-DEP [23]. The resulting diffusion profile takes the form of an error function. For fluences 2×10^{17} Fe/cm², the resulting diffusion length in Al₂O₃ is about 100 Å so that the resulting concentration profile is significant in the first 200 Å but nearly vanishes beyond 400 Å.

In contrast to the diffusion approach, binary-collision calculations were made using the code MARLOWE [24]. The advantage of this method is that the ion energies resulting from high momentum transfer between the primary (Fe) ions and secondary (Cr or Zr) ions can be calculated. Using stopping powers in Al₂O₃ estimated by E-DEP [23], we approximated the ranges of these primary and secondary ions making a small angular correction for "projected" range. The results are shown in Figs. 1 and 2 where a superposition of Cr and Fe are shown (as previously mentioned, the measured profile does not distinguish between Cr and Fe or Zr and Fe at large Zr or Cr depths). The deeper regions (≈ 2000 Å) consist mostly of Fe ions degraded by collisions in the Cr film while the shallow regions (≈ 1000 Å) are mostly Cr atoms. The nearly featureless region (1000-2000 Å) is the overlap between the low energy Fe "tail" and the high energy Cr "tail". The corresponding Zr + Fe curve in Fig. 2 is qualitatively similar.

Although neither the diffusion curve alone nor the binary collision curve alone fits the data well, there is reason to believe that a coherent addition of the theories may explain the data. If we add to the Fe component the energy degradation of the primary beam by the substrate, the deeper regions will more closely agree with the data. The diffusion equations should contain a generator source term, $G_r(x,t)$, given by the Cr or Zr transmission or recoil profile of the binary collision calculations:

$$\frac{\partial C(x,t)}{\partial t} = D \frac{\partial^2 C}{\partial x^2} (x,t) + G_r(x,t). \quad (2)$$

The Knoop impression depths (≈ 3000 Å) exceed the depth of the high (ion mixed) metal concentration region. As a result, the relative hardness values indicate the direction of property changes, but they may underestimate the magnitude of the effects in the implanted region. Nevertheless, this data indicates that the Fe⁺ ion beam itself made a major contribution to the surface

hardness. It is not clear if the hardness in this case is from the radiation damage caused by the beam or due to the implanted ions themselves, although evidence supports the latter. The channeling measurements indicated only slight surface disorder caused by the Fe fluence. The range of the Fe ions is about 4000 Å while the Knoop indentation is only slightly less, about 3000 Å, and probably senses a somewhat deeper surface region. The Cr and Zr presence in the surface layers increased the hardness. The Zr atoms, while showing a lower degree of substitutionality, manifested a greater effect toward increasing the surface hardness.

SUMMARY

We have used the ion beam mixing technique to prepare Cr and Zr enriched surfaces on Al₂O₃. Ions of 1 MeV Fe⁺ were used to irradiate the evaporated metal coatings on Al₂O₃ producing metal concentrations up to a few atomic percent. Channeling measurements suggest that most of the Cr or Zr atoms migrated into substitutional sites. The surface disorder in the Al₂O₃ substrate appeared minimal even for an ion beam fluence of 2×10^{17} Fe/cm². Binary collision calculations were used to investigate recoil effects in this high energy ion beam mixing. Knoop microhardness tests were made on the irradiated surfaces. The results showed significant increases in surface hardness compared to unirradiated surfaces.

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